

GEL'MAN, L. I., Cand Tech Sci -- (diss) "Study of heat exchange  
in <sup>the</sup> drop concentration of mercur<sup>y</sup> vapor." Len, 1958, 11 pp  
(Len Polytechnic Inst im M.I. Kalinin) 100 copies  
(KL, 23-58, 105)

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GEL'MAN, L.I.

AUTHOR: Gel'man, L.I. (Engineer)

96-3-13/26

TITLE: Heat exchange during dropwise condensation of mercury vapor.  
(Teploobmen pri kapel'noy kondensatsii rtutnogo para.)

PERIODICAL: Teploenergetika, 1958, No.3. pp.47-50 (USSR).

ABSTRACT: Knowledge of heat exchange during condensation of mercury vapour is of value in the design of heat exchange equipment in power installations using mercury as a working fluid. The mechanism of the process is also of interest from the general standpoint of the condensation of metal vapours. However, little has been published on this subject at home and abroad. An experimental set-up to study the problem is described and illustrated in Fig.1. The mercury vapour generator is heated by electric radiation furnaces. The output is 150 - 170 kg/hr of mercury vapour, which passes through a throttle valve into the experimental condensor. Numerous refinements are provided on the circuit. The experimental mercury condensor consists of an internal tube of 17/8 mm diameter through which cooling water is passed, and an external tube of 42/34 mm diameter. Mercury vapour is fed to the outer tube and is condensed on the surface of the inner tube. The usual temperature measuring arrangements are provided. An experimental outer tube of heat resistant glass was constructed to permit visual observation and high speed cinematography of the process of condensation of mercury vapour. The mechanism of

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Heat exchange during dropwise condensation of mercury vapor.

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condensation of mercury vapour is then described. Dropwise condensation was to be expected because mercury does not wet a steel surface. Condensation was indeed dropwise and the enlargement of a single frame of the kine film shown in Fig.2. gives some idea of the process. The theory of heat transfer during dropwise condensation is briefly discussed and the process of drop formation is described. The results of the experimental investigation are then given. The relation between the heat transfer coefficient and the temperature head obtained during condensation of mercury vapour on vertical and horizontal tubes is given in Fig.3. The results show that the orientation of the condensation surface has practically no influence on the rate of heat exchange. Fig.4. shows the relationship between the heat transfer coefficient and the temperature head for mercury vapour pressures of 0.8 and 0.15 atm. The data of Figs.3 & 4 show that the heat transfer coefficient is inversely proportional to the temperature head and rises with increase in the mercury vapour pressure. A compound graph of the relationship between the rate of heat flow and the pressure for various velocities of mercury vapour is given in Fig.5 and an empirical formula is given. Fig.6 is a combined graph for the whole of the results from which are derived two formulae that are recommended for use during calculations of heat transfer involving dropwise condensation of pure mercury vapour.

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Additional tests were made to investigate the influence of the presence of air on the intensity of heat exchange during condensation of vapour from a mercury air mixture. The apparatus and procedure were as before. It was found from the experimental results that the influence of the partial pressure of the mercury vapour and of the velocity of mercury air mixture were numerically about the same as during the condensation of pure mercury vapour. The results of the tests with air contents up to 12% are given in Fig.7, the dotted line shows the intensity of heat exchange for mercury vapour not containing air. If the concentration of air by weight is up to 1% heat transfer is not impaired. When this concentration is greater than 1%, the graph given in Fig.8. shows that the experimental points lie satisfactorily around a straight line and corresponding formulae are given for heat transfer calculations involving condensation of mercury vapour from a mercury air mixture. There are 8 figures, 1 literature reference (Russian).

ASSOCIATION: Central Boiler and Turbine Institute (Tsentral'nyy Kotloturbinnyy Institut)  
AVAILABLE: Library of Congress.

Card 3/3

GEL'MAN, L.I., inzh.; KORNEYEV, M.I., kand.tekhn.nauk

High-pressure marine steam generators. Sudostroenie 24 no.4:59-63  
Ap '58. (MIRA 11:4)  
(Marine engines)

ARSLANOVA, A.Kh.; BELYAKOV, V.D.; BERGER, B.I.; VASIL'YEV, A.S.; GAVRILOV,  
N.A.; GEL'MAN, L.I.; KALUGIN, V.P.; KOROSTELEV, V.Ye.; KHAMER,  
I.I.; MIKHAYLOVSKIY, V.T.; ROGOZIN, I.I.; SEREBRYAKOV, L.V.

Combined vaccination with chemical and living vaccines. Voen.-med.  
zhur. no. 1:78-80 Ja '60. (MIRA 14:2)  
(VACCINATION)

3504/3  
S/693/61/000/000/006/007  
D203/D302

26.5500

AUTHOR: Gel'man, L.I.  
TITLE: Experimental investigation of heat transfer in condensation of mercury vapor  
SOURCE: Kutateladze, S.S. ed., Voprosy teplootdachi i gidravliki dvukhfaznykh sred; Sbornik statey, Moscow, Gosenergoizdat, 1961, 156-177

TEXT: The test rig consisted of an electrically heated vapor generator, three condensers, valves and a cooling water system with two electric heaters and two measuring tanks. Vapor was first passed to the experimental condenser, in which cooling water flowed through the inner steel tube. Temperature of the cooling surface was obtained by adding a calculated correction to the readings of two thermocouples built into the wall of the tube. A special technique was developed for fitting these thermocouples. Temperature and pressure of the vapor around the tube was also recorded. The mean velocity of the vapor  $w$  was calculated from the mean flow.

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Experimental investigation of ...

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End heat losses did not exceed 0.5% and the general theory of errors was used to assess the accuracy of the tests. There is no full theory for drop condensation of mercury. The theoretical formula of S.S. Kutateladze is approximate and valid only for vapor without motion. The process was filmed while the instrument recordings were taken at the same time to tie up the qualitative characteristics with the numerical data. The droplets usually began to move before reaching their breakaway size. Due to this motion, only 3.6% of the surface was covered by visible drops. The number of centers of condensation (droplets) was about  $5.2 \times 10^5$  per  $m^2$ . This number and the frequency of droplet formation were practically independent of the heat load, but the projected area of the drops increased with the heat load. The coefficient of heat transfer was found to be inversely proportional to the temperature difference,  $\Delta t$ , between the vapor and the cooling surface, and increased with the vapor pressure  $p$ . It was found that Eq.

$$\frac{\alpha}{p^{1/3} [1 + (\gamma/\omega)^{1/3}]} = f(\Delta t)$$

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Experimental investigation of ...

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and that the heat load  $q$  can be given by Eq.(5)

$$q = 1,2 \cdot 10^5 p^{1/3} [1 + (\gamma w)^{1/3}] [kka M^2 q] \quad \text{for all the 150}$$

test points. Percentage of air by weight  $\xi$  above 1% produced a reduction of heat transfer according to formula Eq.

$$\frac{\alpha_{\xi 0,2}}{p^{1/3} [1 + (\gamma w)^{1/3}]} = f(\Delta t) \quad \text{which also holds for pure mercury}$$

vapor if  $\xi$  is put equal to unity. About 0.05% of magnesium by weight is needed to intensify heat transfer into boiling mercury in a boiler. Tests showed that only traces of Hg are carried away with the vapor. These are usually 20 to 40 times less than the minimum amount to produce wetting of the cooling surface. Thus droplet condensation with its higher coefficient of heat transfer is always ensured. There are 5 tables, 18 figures and 7 Sovietbloc references.

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X

S/096/63/000/004/005/010  
E194/E455

AUTHORS: Gell'man, I.I., Candidate of Technical Sciences,  
Kolosov, V.V., Candidate of Technical Sciences,  
Tyul'nev, I.I., Engineer

TITLE: Heat circuits of binary mercury-water nuclear power  
stations

PERIODICAL: Teploenergetika, no.4, 1963, 49-52

TEXT: The binary mercury-steam cycle promises higher thermal efficiency of nuclear power stations, although mercury can only be used directly in a fast neutron reactor: in other types an additional heat-transfer medium is required. A thermal block diagram is suggested of a power station with an output of 180 MW. Of this, the mercury set working at an evaporation rate of 4015 t/hour generates 80 MW; the steam set generates 100 MW with steam conditions of 35 atm, and 455°C, obtained by a combination of cooling water from the mercury condenser and feed-water heating from the mercury turbine. Because of the interdependence of the mercury and steam circuit conditions it is quite a complicated matter to select the optimum cycle. The overall thermal  
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Heat circuits of binary ...

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E194/E455

efficiency is affected by the number of steam superheaters and on the positions from which the mercury vapor is tapped to heat them. This problem is investigated theoretically by formulating a balance of the work that can be obtained from the cycle, allowing for the quantity of heat used. Comparisons can then be made between equipments with various numbers of super-heaters, and the best positions of the tapping points determined. By way of example, a binary cycle is considered with a steam turbine of 100 MW, steam conditions of 90 atm, 535°C, feed-water temperature 220°C, and mercury vapor at 236 atm, 600°C, with a pressure of 0.6 atm in the mercury condenser. The use of additional mercury superheaters gives diminishing advantages and their number should not exceed 3. Indeed, the transition from two to three superheaters increases the overall efficiency by less than 1% and considerably complicates the heat circuit, so that the best number of steam superheaters is 2. The first tapping point should be in the penultimate stage of the turbine; the second should be in the stage whose mercury vapor conditions are such that the steam can be heated to the required temperature. In this case the

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Heat circuits of binary ...

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efficiency of the mercury part of the installation is about 7% higher than in the case of single stage superheat. A factor which limits the potential use of mercury in nuclear power stations is the low critical heat flux which, for magnesium amalgams is of the order of  $4 \times 10^5$  kcal/m<sup>2</sup>hour. Further experimental work is required for solving the problem of intensifying heat exchange of boiling mercury. Loadings of  $1.6 \times 10^6$  kcal/m<sup>2</sup>hour have been obtained in the laboratory. The use of a binary mercury/steam cycle can raise the overall efficiency of nuclear power stations to 45 to 51%, which is much higher than the efficiency obtained with other heat-transfer media and so the method is, in principle, promising. There are 3 figures and 1 table.

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L 1050-66 EWT(1)/EPF(c)/ETC/EPF(n)-2/ENG(m) WW/75

ACCESSION NR: AT5016894

AUTHOR: <sup>44.35</sup> Borishanskiy, V. M.; <sup>44.35</sup> Gel'man, L. I.; <sup>57</sup> <sup>B41</sup> Zablotskaya, T. V.; <sup>44.35</sup> Ivashchenko, N. I.; Kopp, I. Z. UR/0000/64/000/000/0350/0362

<sup>44.35</sup> <sup>44.65</sup> TITLE: Investigation of heat transfer during the flow of mercury through horizontal and vertical tubes <sup>21, 44.35</sup>

SOURCE: Konvektivnaya teploperedacha v dvukhfaznom i odnofaznom potokakh (Convective heat transfer in two-phase and single-phase flows). Moscow, Izd-vo Energiya, 1964, 350-362

TOPIC TAGS: mercury, heat transfer, liquid flow, forced flow

ABSTRACT: The transfer of heat to mercury is studied during forced flow in horizontal and vertical tubes. The experimental equipment and procedure are described briefly. The following parameters are measured during the experiments: the rates of flow of the liquid, the power input for heating the working section of the equipment, the temperature of the mercury entering and leaving the working section, the temperature fields at various points through the cross section of the tube, the wall temperature at these points and along the tube, the temperatures within and on

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ACCESSION NR: AT5016894

the surface of the insulation for the working section. The results are tabulated. Experimental and theoretical data show excellent agreement. Heat transfer beyond the section of thermal and hydrodynamic stabilization in the absence of thermal contact resistance for Peclet numbers from  $10^3$  to  $2 \cdot 10^4$  may be calculated from the formula  $Nu = 7.5 + 0.005Pe$ . A relationship is found between thermal contact resistance and Reynolds numbers for a vertical tube. Orig. art. has: 9 figures, 5 formulas, 4 tables.

ASSOCIATION: none

SUBMITTED: 17Nov64

ENCL: 00

SUB CODE: TD, ME

NO REF SOV: 007

OTHER: 004

Card 2/2

DP

GEL'MAN, L.I., kand. tekhn. nauk; GASHLOVSKIY, A.N., inzh.

Power and engineering equipment of a system with a mercury heat carrier. Energomashinostroenie 10 no.6:37-39 Jo '64.  
(MIRA 17:9)

GEL'MAN, I.S.; KISHKO, I.S.

Noterage method for determining the yield of cut articles  
from knitted fabrics. Tekst. prom. 24 no.11:71-72 4 1971.  
(MPP 12:12)

1. Ispolnyayushchiy obyazannosti glavnogo inzhenera Makachevskoy  
trikotazhnoy fabriki (for Gel'man). 2. Glavnyy bukhgaltzer  
Makachevskoy trikotazhnoy fabriki (for Kishko).



L 5275-66 EWT(1)/EPA(s)-2/EWT(m)/EPF(c)/ETC/EPF(n)-2/BWG(m)/EWP(t)/EWP(b) IIP(c)

ACC NR: AP5025683 JD/NW/JG SOURCE CODE: UF/0286/65/000/018/0030/0030

AUTHORS: Kanayev, A. A.; Gel'man, L. I.; Kopp, I. Z. 44,55 68 23

ORG: none

TITLE: A method for intensifying heat exchange during boiling of mercury. Class 17, No. 174643 [announced by Central Scientific Research Boiler and Turbine Institute imeni I. I. Polzunov (Tsentral'nyy nauchno-issledovatel'skiy kotlo-turbinnyy institut)] 44,55 21

SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 18, 1965, 30

TOPIC TAGS: mercury, heat exchange

ABSTRACT: This Author Certificate presents a method for intensifying heat exchange during boiling of mercury. To increase the intensity of heat flow, the heat exchange surface is kept in contact with mercury up to the temperature of 600-800C. This temperature is maintained for over 25 hours.

SUB CODE: TD/ SUBM DATE: 10Aug64/ ORIG REF: 000/ OTH REF: 000

Card 1/1 (80) UDC: 621.565.94:536.248.2:669.79

L 38752-66 ENT(1)/ENT(m)/T/ENT(f) DJ/AV/JW

ACC NR: AP6024818

SOURCE CODE: UR/0096/66/000/008/0043/0047

AUTHOR: Gel'man, L. I. (Candidate of technical sciences ; Smolkin, Yu. V. (Engineer)  
; deceased)

ORG: Central steam turbine institute (Kentralnyy kotloturbinnyy institut)

TITLE: Thermodynamic calculation of a gas turbine unit using an electronic digital computer 2/ 23 58

SOURCE: Teploenergetika, no. 8, 1966, 43-47

TOPIC TAGS: gas turbine, turbine design, closed cycle gas turbine, thermodynamic calculation, entropy

ABSTRACT: A computer method was developed for calculating the optimum thermodynamic design of a closed-cycle gas turbine unit using nitrogen as the working fluid.

Emphasis was placed on the real properties of the gas. Procedures for calculating the final temperature after isentropic expansion and the enthalpy are given. Calculations were made for inlet pressures of 2—10 Mn/m<sup>2</sup>, at a turbine inlet temperature of 1073.16K and a compressor inlet temperature of 298.16K. Plots were obtained for the variation in the internal work as a function of the inlet pressure and expansion ratio, and the variation in the compressor temperature gradient as a function of the turbine inlet pressure. It was shown that at an expansion ratio of 2.5, the internal efficiency at a turbine inlet pressure of 10 Mn/m<sup>2</sup> is 2.81% higher than that of an ideal gas. At a ratio of 5.5, it is only 1.48% higher. It is concluded that an allowance for the real-gas properties at turbine inlet

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UDC: 621.438.681.142.35:001.24

L 38782-66

ACC NR: AP6024818

0

pressures higher than  $2 \text{ Mn/m}^2$  is necessary when nitrogen is used as the working medium. [PV]

SUB CODE: 13,20/SUBM DATE: none/ ORIG REF: 002/ OTH REF: 001/

Card

2/2

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GEL'MAN, I. S.

GEL'MAN, L.S., inzhener.

Remarks on new principles of construction and arrangement of a  
substation. Elek. sta. 25 no.6:59 Je '54. (MLRA 7:7)  
(Electric substation)

BULGARIA/Nuclear Physics - Installations and Instruments.  
Methods of Measurement and Research.

C

Abs Jour : Ref Zhur Fizika, No 12, 1959, 26730  
Author : Gel-Man, M., Rosenbaum, Ye.  
Inst : ~~Elementary Particles~~  
Title : Elementary Particles  
Orig Pub : Fiz.-matem. sprsaniye, 1958, 1, No 3-4, 89-118  
Abstract : See Referat Zhur Fizika, 1958, No 3, 5358.

Card 1/1

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GEL'MAN, M.I.; BIRANIN, V.G.; BELYAYEVSKIY, A.G.; ANDREYEV, A.I.;  
BELYAYEV, V.P.; PETROV, V.I.

On new technological processes. Der.prom.4 no.1:19-21 Ja'55.  
(MLRA 8:3)

1. Ust'-Ishorskiy faseruyy zavod.  
(Ust'-Ishora—Plywood)

GEL'MAN, M.I., inzh.; TUPITSYN, Yu.S., inzh.; EL'BERT, A.A., inzh.

Bartrev's method for manufacturing hardboards of wood shavings.  
Der. prem. 8 no.7:25-26 JI '59. (MIRA 12:9)

1. Ust'-Izherskiy fanernyy saved.  
(Hardboard)

BOTVINIK, Yefim Solomonovich; IMITRIYEV, Oleg Aleksandrovich; GEL'MAN, Moisey Isaakovich; TUPITSIN, Yuriy Semenovich; EL'BERT, Aleksandr Aronovich; VARAKSIN, F.D., red.; LEBEDEV, I.D., red. izd-va; PARAKHINA, N.L., tekhn. red.

[Use of the continuous method for the manufacture of particle boards]Proizvodstvo struzhechnykh plit nepreryvnym sposobom. Moskva, Goslesbumizdat, 1961. 98 p. (MIRA 15:2)  
(Hardboard) (Assembly-line methods)



GEL'MAN, M.L.

Recent find of gedrite in the U.S.S.R. Dokl. AN SSSR 141  
no.3:709-712 N '61. (MIRA 14:11)

1. Severo-Vostochnoye geologicheskoye upravleniye. Predstavleno  
akadomikom D.S. Korzhinskim.  
(Lesser Anyuy Valley—Gedrite)

GEL'MAN, M.L.

Reflection of the microheterogeneity of the magmatic melts in the structure of augen-diorite enriched with titanium. Geokhimiia no.2:147-153 '62. (MIRA 15:3)

1. North-Eastern Geological Department, Magadan.  
(Magma) (Titanium) (Diorite)

GEL'MAN, M.L.

Triassic diabase formation in the Anyuy Stone (Chukchi National Area).  
Geol. i geofiz. no.2:127-134 '63. (MIRA 16:5)

1. Severo-Vostochnoye geologicheskoye upravleniye, Magadan.  
(Chukchi National Area—Diabase)

GEL'MAN, M.L.

Relation between igneous activity and granitoid intrusions  
in the western Chukchi Peninsula. Izv. AN SSSR. Ser. geol.  
28 no.12:41-58 D'63. (MIRA 17:2)

1. Severo-Vostochnoye geologicheskoye upravleniye, Magadan.

GEL'MAN, M.L.

Plutonic facies and formation phases of the granitoid complex of  
the Anyuy zone. Dokl. AN SSSR 149 no.6:1397-1400 Ap '63.  
(MIRA 16:7)

1. Severo-vostochnoye geologicheskoye upravleniye. Predstavleno  
akademikom D.I.Shcherbakovym.

(Chukchi Peninsula--Granite)  
(Chukchi Peninsula--Geology, Stratigraphic)

*GELMAN, M.M.*

C

(2) 1150

Metal/muffle small size tunnel kiln. M. M. GELMAN, *Nirbol-nye i Kram. Prom.*, 8 [6] 17-19 (1947).—One of the main reasons why such kilns have not found wide acceptance in the Soviet Union was the apprehension that the production of fire-clay muffles would be unusually complicated and that they would require frequent repairs. In 1937, an 8.5 x 3.12 x 2.25 m. kiln for firing plates was constructed at the Riga Tile Works. The kiln has a preheating, a firing, and a cooling zone. Under the firing zone is a three-stage smogas drebok and the recuperators. Along the length of the kiln are three rows of muffles; in each row are 8 muffles, one on the other. The inner cross section of the muffles is 60 x 230 mm. The muffles consist of individual snugly positioned rings 310 and 360 mm. long. The hourly output is 120 plates. The kiln operated 4 yr. without requiring a change in muffles. 3 illustrations. B.Z.K.

454-SL METALLURGICAL LITERATURE CLASSIFICATION

FROM SOURCE SEARCHED MAP ONLY COT COLLECTIONS

U S A M I S R P O N T W H S A C J V

MA MA

AGALAROV, Ch.S.; ALESKEROV, S.A.; GUL'MAN, M.M.; GINZBURG, M.Ya.; IRRAGIMOV,  
I.S.; ZUL'FUGARZADE, E.; MAMEDLI, E.M.

"Information converter for electronic digital computers" by E.I.  
Gitis. Reviewed by Ch.S. Agalarov and others. Izv.tekh. no.7:  
64 J1 '62. (MIRA 15:6)  
(Electronic digital computers)  
(Gitis, E.I.)

45642

S/877/62/001/000/004/003  
D201/D308

9,7500

AUTHORS: Alekserov, S.A., Gel'man, M.M. and Kasumov, R.Ya.  
TITLE: A fast generator-counter system  
SOURCE: Akademiya nauk Azerbaydzhanskoy SSR. Vychislitel'nyy  
tsentr. Trudy, v. 1, 1962, 38-45

TEXT: The authors describe the circuits and the operation of a nanosecond pulse generator and an associated binary counter. The pulse generator consists of a crystal controlled oscillator, buffer stage, used also as a suppressor-controlled gate, limiter and inductive differentiating stage and finally a pulse-shaping output stage. All stages have pulse-transformer coupling. Pulses of nanosecond duration are obtained from heavily damped transients in the pulse transformer of the differentiating stage and by diode loading of the output stage. Ferrite cores are used throughout. The output pulse amplitude is about 20 v, repetition frequency of the order of 8 Mc/s, pulse duration 0.04  $\mu$ sec. The binary counter following the pulse generator consists of two flip-flops, the first with H<sup>+</sup> anode

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A fast generator-counter system

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D201/D308

circuit correction, separated by pulse amplifying stages. The circuit utilizes valves with small stray and intelectrode capacitances. The first flip-flop operates at pulse repetition frequencies up to 5 Mc/s; the second flip-flop at up to 2 Mc/s, with output pulse amplitudes of about 60 v. The carry pulse is obtained by RC differentiation, amplitude about 15 v, duration between 0.04 and 10 microseconds. Tolerance of components is  $\pm 20\%$ . The above generator counter system may be used in time-modulator digital-analog and analog digital converters. There are 9 figures. X

2/2

ALESKEROV, S.A.; GEL'MAN, M.M.; KASUMOV, R.Ya.

A high-speed generator-counter system. Trudy Vych.  
tsentra AN Azerb. SSR 1:38-45 '62. (MIRA 15:11)  
(Radio measurements)  
(Pulse techniques (Electronics))

L 19679-65 EWC(d)/EED-2/EWP(1) Po-l/Pq-l/Pg-l/Pk-l IAP(c)/AEDC(a)/SSD/BSA/AFWL/  
ASD(a)-5/ASD(c)/AEDC(d)/AFMDC/AFETR/PAEM(a)/AFTC(b)/RAEM(d)/ESD(c)/ESD(dp) GG/BB  
ACCESSION NR: AP4038886 S/0119/64/000/005/0012/0013

AUTHOR: Abrosimov, I. L.; Aleskerov, S. A.; Akhundov, E. I.;  
Gel'man, M. M.

TITLE: Semiconductor analog-to-digital voltage converter 160 B

SOURCE: Priborostroyeniye, no. 5, 1964, 12-13

TOPIC TAGS: automatic control, industrial automatic control, analog digital  
converter, digital computer, semiconductor analog digital converter

ABSTRACT: A new voltage-to-code converter is intended for introducing  
process-sensor information into a digital computer for the purpose of centralizing  
supervision and control of the process. The well-known principle of comparing  
the input voltage with a linearly-variable voltage is used; the input variable is  
converted into a time interval. The linearly-variable voltage is obtained by  
integrating a square pulse; a square-pulse shaper and a d.c. amplifier perform

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L 19679-55

ACCESSION NR: AP4038886

this operation. A transistorized comparison device yields the time intervals proportional to the running value of the input voltage. A special transistorized gate is controlled by the comparator pulses and turns a pulse generator on and off. The latter produces 5-v, 0.25-microsec pulses at a repetition frequency of 1 mc. The number of pulses equivalent to an input voltage value is counted by a transistorized binary counter. Max input voltage, 20 v; conversion frequency, 300 cps; ambient temperature, up to 40C; claimed apparatus error, 0.2%.  
Orig. art. has: 4 figures.

ASSOCIATION: none

SUBMITTED: 00

ENCL: 00

SUB CODE: DP, EC

NO REF SOV: 003

OTHER: 000

Card 2/2

GEL'MAN, M.M.; TAVROVSKIY, A.D.

At the Japanese Industrial Exhibition in Moscow. Izv. tekhn.  
no.10:38-43 0 '65. (MIRA 18:12)

8(2)

AUTHORS: Fleyshman, L. S., Engineer, Gol'dnan, SOV/105-58-11-10/28  
M.V., Engineer

TITLE: Investigation of Inverter Duty of Type RIV-500 x 6  
(Issledovaniye invertornogo rezhima vypryamiteley  
RIV-500 x 6)

PERIODICAL: Elektrichestvo, 1958, Nr 11, pp 43 - 47 (USSR)

ABSTRACT: In this paper the causes for an unstable operation of  
an inverter rectifier are exposed. The investigations were  
carried out in the Laboratory for Mercury-Arc  
Rectifiers of the "Uralslektroapparat" plant. This  
paper also includes results of the tests which were  
made with special measures for increasing the reliability  
of the inverter mode of operation. When making a  
choice between different circuit conditions of an  
inverter unit, the following three circuits come into  
the picture : 1) Delta, six phase, double way. This  
circuit was tested under operational conditions on the  
Yuzhno-Ural'skaya and Sverdlovskaya zheleznaya doroga  
(Sverdlovsk Railroad). 2) Three-phase diametric double

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Investigation of Inverter Duty of Type RMTV-500 X 6

SOV/105-58-11-10/28

way. This was tested on the test stands of the plant laboratory and on the Sverdlovsk railroad. 3) Delta, six phase fork. This circuit is almost exclusively used abroad (Ref 1) in inverter units. R.B.Gafirov, Z. Kh. Chernin and Ye.V.Libina, Engineers at the Laboratory for Mercury-Arc Rectifiers of the "Ural-elektroapparat" plant, assisted in the work. The experimental array is described. Causes for arc-through are as follows: A too short period for the regeneration of the controlling capability of the grid. 2) Arc-back. 3) Extinction of the excitation. 4) Inductance in the grid circuit. 5) Loss of control during voltage rise at the valve. The majority of arc-throughs in a three-phase diametric double way circuit were recorded for the moment of ignition of the inverse phase valve. The cause of such arc-throughs is found in the rapid rise of the direct voltage when the de-ionization is not yet completed. For this reason the test stand circuits (which are intended for checking the valves for an inverter operation) must be in a

Card 2/4

Investigation of Inverter Duty of Type **RMIV-500 x 6<sub>k</sub>**      SSV/105-58-11-10/28

position to supply this positive peak. The absence of the cathode spot in ignitrons during the non-conducting period permits to draw the conclusion that for ignitrons operating in an inverter regime the loss of control capability during the direct voltage rise is not dangerous, this fact indicating their suitability for such a mode of operation. The investigations lead to the following conclusions: 1) The occurrence of a considerable number of arc-throughs at the ignition of the inverse-phase valve made necessary a check of the requirements placed upon the test stand circuits. 2) An establishment of circuits shunting the valve and of a reactor coil in the cathode branch with an inductivity of 50 to 100 mH considerably increases the reliability of the **RMIV-500 x 6** rectifier in an inverter mode of operation. 3) The load level attained (500 A continuously, 700 A for 15 minutes, and 800 A for 10 minutes) guarantees a regenerative braking operation of the rectifier. 4) The results of the test runs of the inverter enabled the plant to construct three test inverter units for the substations Goytkh and Tverskaya

Card 3/4



Investigation of Inverter Duty of Type **RMIV** -900 x 6

SOV/105-9-11-10/28

of the Severo-Kavkazskaya zheleznaya doroga (North Caucasus Railway) and the substation Neyvo-Rudyanka of the Sverdlovskaya zheleznaya doroga (Sverdlovsk Railway). The investigation was carried out due to the initiative of Ye.M.Glukh, Candidate of Technical Sciences. There are 8 figures and 5 references, which are Soviet.

ASSOCIATION: Zavod "Uralelektroapparat" (Plant "Uralelektroapparat")

SUBMITTED: May 14, 1957

Card 4/4

GEL'MAN, M.V., inzh.

Increase in the stability of the excitation of mercury  
rectifiers using semiconductor ignitrons. Vest. elektroprom.  
34 no.2:17-21 F '63. (MIRA 16:2)  
(Mercury-arc rectifiers)  
(Electric railroads—Current supply)

AKODIS, Mikhail Mironovich dr. tekhn. nauk, prof. GEL'MAN, Boris Vladimirovich, aspirant

Study of the grid circuit of a multistage frequency converter.  
Izv. vys. ucheb. zav. elektromekh. 7 no. 4:428-435 '64  
(MIRA 17:7)

1. Kafedra tekhniki vysokikh napryazheniy Uralskogo politekhnicheskogo instituta. 2. Zaveduyushchiy kafedroy tekhniki vysokikh napryazheniy Ural'skogo politekhnicheskogo instituta (for Akodis).

L 22186-66 EWA(h)/EWT(1)

ACC NR: AP6012959

SOURCE CODE: UR/0143/65/000/003/0014/0022

AUTHOR: Akodis, M. M. (Doctor of technical sciences; Professor); Gell'man, M. V.  
(Engineer)

ORG: Ural Polytechnic Institute im. S. M. Kirov (Ural'skiy politekhnicheskiy  
institut)

TITLE: Automatic control of a sequential frequency converter 25

SOURCE: Izvestiya vysshikh uchebnykh zavedeniy. Energetika, no. 3, 1965, 14-22

TOPIC TAGS: automatic control, electronic circuit, frequency converter, electronic  
rectifier, electric resistance, electric inductance, electric capacitance

ABSTRACT: The possibilities of automatic control of ion or semiconductor  
sequential inverters, so that they may be used for technological heat pro-  
cesses, is analyzed. An approximate method is developed for design of a  
sequential inverter, loaded with a parallel oscillating circuit. Control  
criteria are analyzed, with the goal of keeping the operation of the inverter  
constant with variation in the load. Contactless operation, most simply  
achieved by changing the control frequency, is seen to be preferable to con-  
tact control by switching of compensating capacitances. A phase sensitive  
rectifier can be used as a transducer in controlling the frequency of the  
inverter, in order to keep it in resonance with the frequency of the load  
circuit. This type of control is most suitable where there are only slight  
variations of the ratio of load resistance to inductance in normal operation.

Cord 1/2

UDC: 621.314.26-523.2

L 22186-66

ACC NR: AP6012959

Where variations in  $r_1/L_1$  are greater, control can be achieved better by using constancy of inverter input current or of voltage at the commutating capacitance, which also leads to constancy of power in the load circuit, as a control criterion. Orig. art. has: 6 figures, 22 formulas, and 1 table. [JPRS]

SUB CODE: 09 / SUBM DATE: 27Apr64 / ORIG REF: 003 / OTH REF: 001

Card 2/2 . nat

AKODIS, M.M., doktor tekhn. nauk, prof.; GEL'MAN, M.V., inzh.

Use of regulated silicon rectifiers in ultrasonic frequency  
converter networks. Elektrichestvo no.3:26-30 Mr '65.

(MIRA 18:6)

1. Ural'skiy politekhnicheskii institut imeni Kirova.

L 07069-67 ENT(1)

ACC NR: AP6019234

(N)

SOURCE CODE: UR/0144/66/000/002/0223/0225

AUTHOR: Gel'man, M. V.; Mineyev, V. A.

ORG: None

TITLE: Investigation of semiconductor master oscillator for a three-cell series inverter 25

SOURCE: IVUZ. Elektromekhanika, no. 2, 1966, 223-225

TOPIC TAGS: semiconductor device, frequency control, ion current, electric current, transistorized oscillator, frequency converter

ABSTRACT: An investigation of a master oscillator, the master stage of which consisted of a three phase semiconductor converter with one master cell and two slave cells, was made. Slave cells were synchronized by feeding part of the collector coil voltage of the master cell to the collector coil circuit of the slave cell transformers. The moment of saturation of the transformer cores can be varied and any phase shift can be produced, regardless of the feed voltage. The frequency of oscillations produced is linearly dependent on the feed voltage over a rather wide range. Rapid frequency control is achieved by connecting a control semiconductor triode into the feed circuit of the master stage. Individual control of the electron and ion currents flowing in the grid circuit of the inverter gate is possible. The master oscillator was used in a three-cell frequency converter circuit. Two variants were made,

Card 1/2

UDC: 621.314.6.+621.501.

L 07069-67

ACC NR: AP6019234

designed for 2,500 and 8,000 cycle per second operation. Frequency control of + 25% of designed frequency was provided for in both cases. The investigation demonstrated that the master oscillator could provide independent frequency and control pulse duration regulation while ensuring a high degree of ignition precision and improved conditions for deionization of the gate. Orig. art. has: 5 formulas, 2 figures and 1 table.

SUB CODE: 09/SUBM DATE: 24Jan64/ORIG REF: 002/OTH REF: 001

Card 2/2 *LC*



ACC NR: AT6022766

(A)

SOURCE CODE: UR/2563/65/000/258/0161/0171

AUTHOR: Gel'man, M. Z.; Ryabov, B. M.

ORG: none

TITLE: Ionization characteristics of polymerized insulation

SOURCE: Leningrad. Politekhicheskiy institut. Trudy, no. 258, 1965.

Vysokovol'tnaya izolyatsiya liniy i apparatov (High voltage insulation of lines and apparatus), 161-171

TOPIC TAGS: electric insulation, electric discharge ionization, polymer dielectric

ABSTRACT: On the basis of well-known J. Berks, J. Shulman, and S. Whitehead works, it is theoretically found that the initial ionization voltage (IIV) in a polymerized insulating material can be raised by impregnating the insulation with a high-electric-strength gas, by the use of a higher vacuum in drying and pouring the insulation, and other processing techniques. A theoretical relation between IIV, the dielectric constant, and the gas-inclusion (bubble) size explains the ionization-voltage values of 2-10 kv often observed in practice. A further analysis shows that:

Card 1/2

ACC NR: AT6022766

(1) The  $\Delta Q_x$  and ionization intensity  $\Delta Q_x$  in a near-uniform field are determined by the inclusions size, and the thickness and  $\epsilon$  of the solid dielectric; the ionization characteristics are not only the functions of the applied voltage but also are determined by statistical voltage distribution among the ionized inclusions; the relative ionization intensity  $I_r$  is, too, a function of the statistical voltage distribution; (2) The number of discharges per second as a function of applied voltage has a region with  $n \approx (\frac{U}{U_0} - 1)^\alpha$ , where  $U_0$  - voltage across the gas bubble,  $\alpha = 1.3-2.0$ ; (3) The  $I_s(U)$  has a region in which  $I_s \approx (\frac{U}{U} - 1)^\beta$ , where  $\beta = 1.4-2.6$ .

Orig. art. has: 8 figures, 35 formulas, and 2 tables.

SUB CODE: 09 / SUBM DATE: none / ORIG REF: 004 / OTH REF: 001

Card 2/2

11-11-77 001 (1)/111(1)

ACC NR: AN5016460

SOURCE CODE: UR/0124/65/000/012/B000/B000

AUTHOR: Gel'man, N. A.

TITLE: Characteristics of semibounded fan-shaped jets

SOURCE: Ref. zh. Mekhanika, Abs. 12B474

REF SOURCE: Nauchn. raboty in-tov okhrany truda VTSSPS, vyp. 2(34), 1965, 6-11

TOPIC TAGS: turbulent jet, thermodynamics, flow characteristic

ABSTRACT: Data are given from an experimental investigation of nonisothermal semi-bounded fan-shaped turbulent jets. The velocity profiles and temperature profiles in the transverse cross sections of the jet are determined as well as the variation in the maximum velocity and maximum excess temperature along the radius. The profiles of relative velocities  $W$  and excess temperature  $v$  as a function of  $y/x$  are practically universal. The relationship between these quantities is  $\sqrt{v}/W$  which is characteristic for free jets. Approximate formulas are proposed for calculating these jets which are strictly true only for free fan-shaped jets. A. S. Ginevskiy. [Translation of abstract]

SUB CODE: 20

Card 1/1

GLUKHAREV, A.I., inzh. (Engel's); FOYGEL', L.A. (Engel's); GEL'MAN,  
H.B., inzh. (Engel's)

Calculation of current in an R-L circuit with half-wave  
rectification. Elektrichestvo no.5:58-60 My '60.

(MIRA 13:9)

(Electric current rectifiers)

(Electronic circuits)

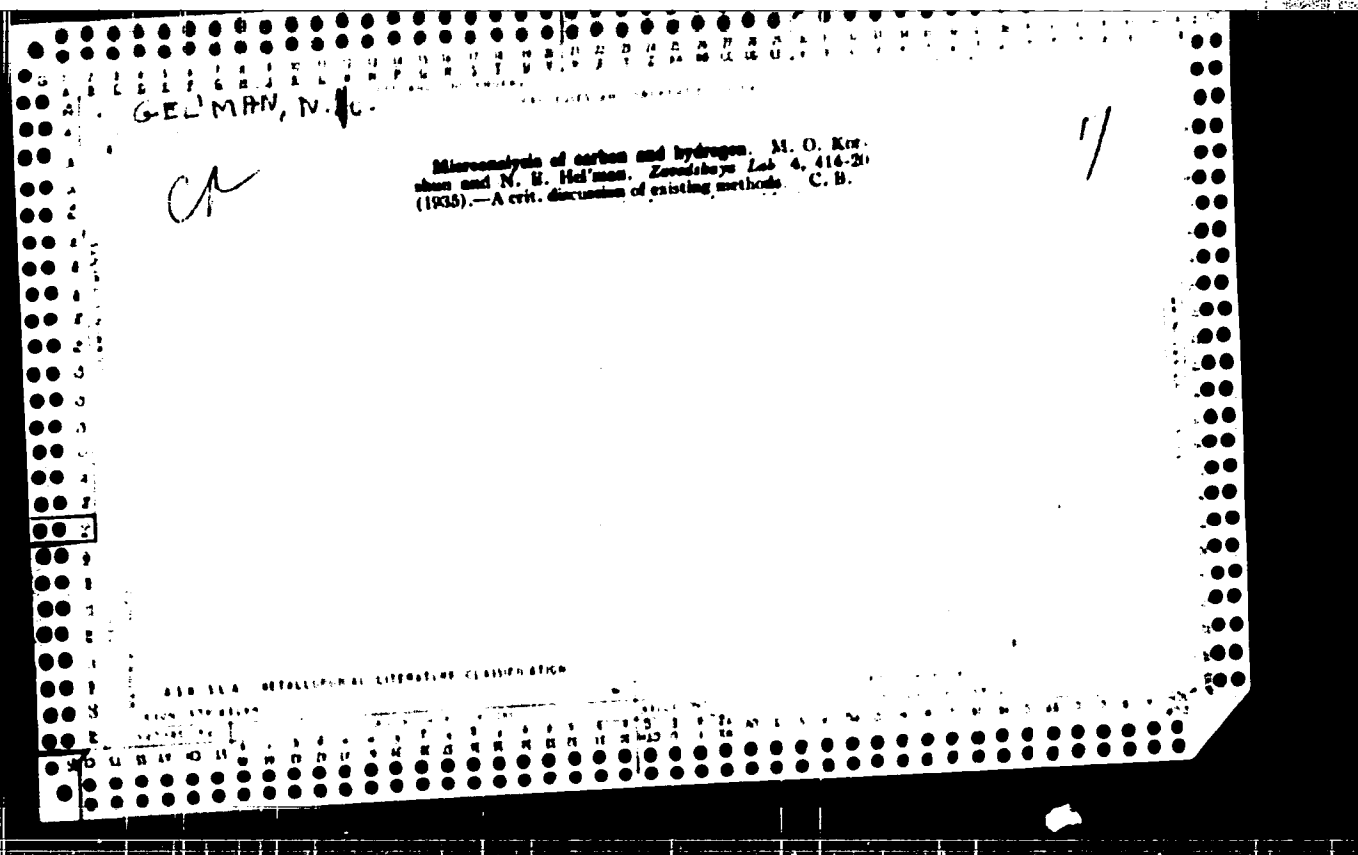
GEL'MAN, N

S

Biokhimiya Rasteniy; Bibliograficheskiy Ukazatel'  
Otechestvennoy Literatury, 1738-1952. Sost.:  
N. S. Gel'man I G. D. Zenkevich. Moskva, Akademkniga,  
1956.

394 P. 27 cm. (Materialy Po Istorii Biologicheskikh Nauk v SSSR).

At head of Title: Akademiya Nauk SSSR. Otdeleniye  
Biologicheskikh Nauk, I Otdel Biobibliografii Uchenykh  
SSSR. Fundamental'noy Biblioteki Obshchestvennykh  
Nauk.

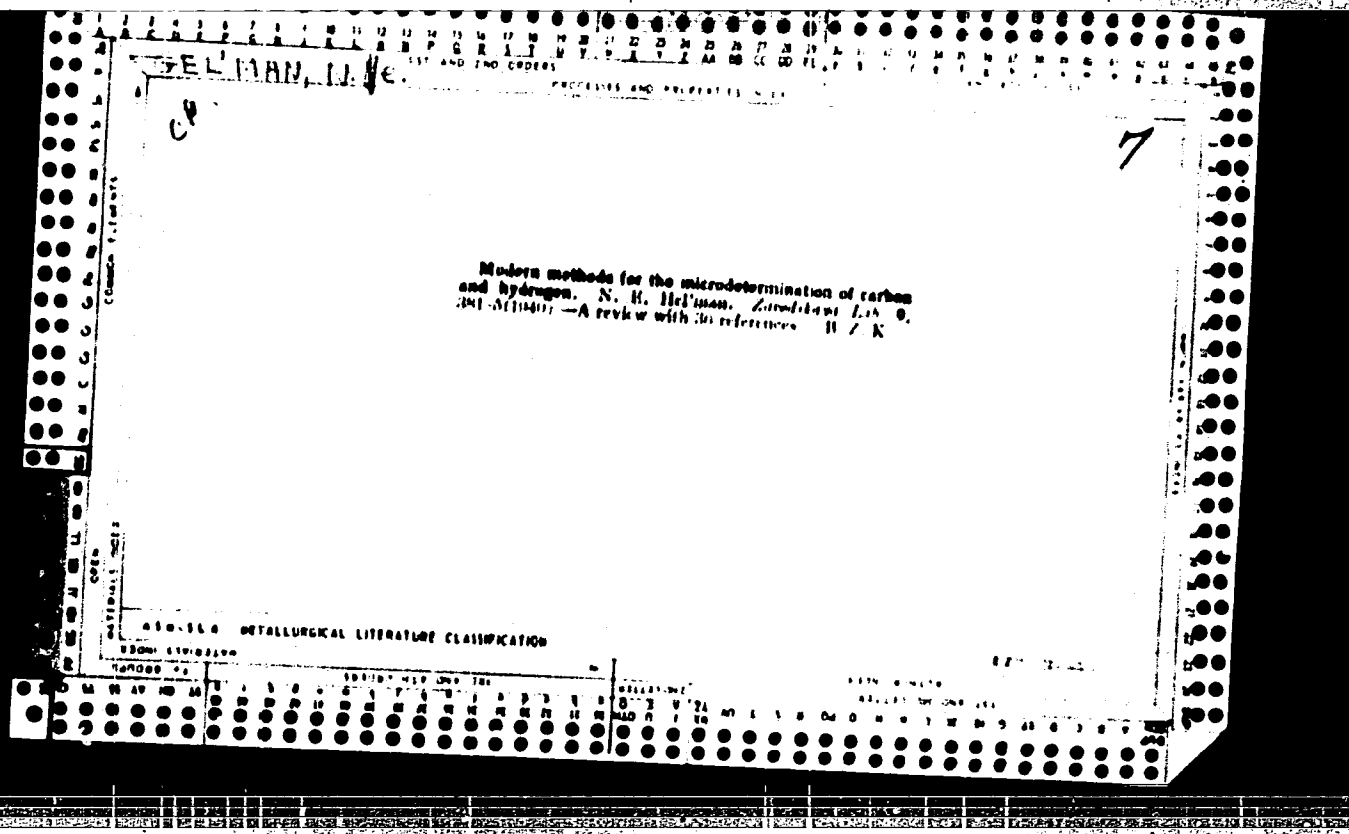


CA

Microdetermination of sulfur in organic compounds  
S. E. Gel'man, *Zashchita Lazh.* 8, 673 (1939). In a modified Meulen method (cf. C. A. 30, 419) a 3.5 mg. sample is heated in a Pt boat, the vapors of the org. compound are passed with a stream of  $H_2$  over a heated catalyst of rolled Pt gauze, the  $H_2S$  is absorbed in a buffered  $ZnSO_4$  soln. and the S is deid. iodometrically. The combustion tube contg. the Pt boat and Pt catalyst is 55 cm. long, 8.9 mm. in diam. (inside), and is made of quartz. The glass absorption tube was bent at  $45^\circ$  with the vertical end connected to the combustion tube and the inclined arm shaped into a series of bulbs. Detns. of S on 8 org. compds. showed close agreement with the theoretical values. The procedure and app. are described in detail. H. Z. K.

850-354 METALLURGICAL LITERATURE CLASSIFICATION

85000 851000 852000 853000 854000 855000 856000 857000 858000 859000 860000 861000 862000 863000 864000 865000 866000 867000 868000 869000 870000 871000 872000 873000 874000 875000 876000 877000 878000 879000 880000 881000 882000 883000 884000 885000 886000 887000 888000 889000 890000 891000 892000 893000 894000 895000 896000 897000 898000 899000 900000 901000 902000 903000 904000 905000 906000 907000 908000 909000 910000 911000 912000 913000 914000 915000 916000 917000 918000 919000 920000 921000 922000 923000 924000 925000 926000 927000 928000 929000 930000 931000 932000 933000 934000 935000 936000 937000 938000 939000 940000 941000 942000 943000 944000 945000 946000 947000 948000 949000 950000 951000 952000 953000 954000 955000 956000 957000 958000 959000 960000 961000 962000 963000 964000 965000 966000 967000 968000 969000 970000 971000 972000 973000 974000 975000 976000 977000 978000 979000 980000 981000 982000 983000 984000 985000 986000 987000 988000 989000 990000 991000 992000 993000 994000 995000 996000 997000 998000 999000 1000000







1ST AND 2ND DEPT'S		PROCESSING AND PROPERTIES INDEX		3RD AND 4TH DEPT'S	
FELMAN, N. C.		CA		1	
<p>An apparatus for quantitative organic microanalysis. I. M. O. Kordun and N. E. Mal'ina. <i>Zashchita Lab.</i> 12, 100-10(1960).—The sample is burned in a current of O in a tube of transparent quartz or of high-melting glass. The combustion products pass over a Pt catalyst and into absorption tubes through a part of the combustion tube filled with metallic Ag and <math>PbO_2</math> (the combustion tube is placed in a thermostat). S, halides, and N oxides are absorbed in this part of the tube. <math>CO_2</math> and water formed in the combustion are absorbed in absorption tubes which can be weighed on a microbalance. Eight references. W. R. Henn</p>					
<p>ASS. I. A. METALLURGICAL LITERATURE CLASSIFICATION</p>					
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100000 #298		100000 #299		100000 #300	

GEL'MAN, N. E.

RT-1020 (Apparatus for direct microdetermination of oxygen. Communication II)  
Apparatura dlia priamogo mikroopredeleniia kislóroda. Soobshchenie II.  
Zavodskaiia Laboratoriia, 12(4-5): 500-502, 1946.

GELMAN, IV. C.										PROCESSING AND PROPERTIES																																																	
CA										1																																																	
<p>An apparatus for microdetermination of sulfur. M. O. Korshak and N. K. Gelman. <i>Zhurnal Khim. Fiz.</i> 12, 754-6(1946).--Vapors of an org. substance are passed in a current of H<sub>2</sub> over a glowing Pt catalyst. The H<sub>2</sub>S formed is absorbed by ZnSO<sub>4</sub> soln. in AcOH and is titrated iodometrically.</p> <p style="text-align: right;">W. R. Hena</p>																																																											
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GELMAN, N. E.

*Chim.* 3 7

*Korshun, M. O., and Gel'man, N. E.: Novye Metody  
Elementarnogo Mikroanaliza (New Methods of Ele-  
mentary Microanalysis). Moscow-Leningrad: Gos-  
khimizdat, 1949. R7 Kop. 60. Reviewed in Uspekhi  
Khim. 18, 375(1949).*

*MA*

[illegible]

8771

A NEW METHOD FOR THE SIMULTANEOUS MICRO-  
DETERMINATION OF FLUORINE, HYDROGEN, AND  
CARBON IN ORGANIC COMPOUNDS.

N. I. Gelfand and  
M. O. Korshyn. Translated from Dokl. Akad. Nauk  
S.S.R. 69, 648-7(1963). 3p. Available from Associated  
Technical Services (Trans. RI-120), East Orange, N. J.  
(AEC-17-1430)

The method consists of burning the sample in a stream  
of O<sub>2</sub> or air in a quartz tube. H and C are determined in  
the usual manner, and F is combined with a metal oxide  
in the combustion tube. The amount of F in the sample is

determined by the change in weight of the oxide. Examples  
are given. (T.S.R.)

Gel'man, h. E.

Simultaneous microdetermination of fluorine, carbon, and hydrogen in heteroorganic compounds V. S. O. Gorshun, N. I. Gel'man, and E. I. Glazova, Doklady Akad. Nauk S.S.S.R. 111, 1255-6 (1956); cf. C.A. 47, 3524h.  
The sample (4-10 mg.) is burned in a stream of O<sub>2</sub> in a layer (16-18 cm. long) of granulated MgO kept at 1000° and held in a quartz cartridge with the perforated end placed

in the empty combustion tube. The C and H detn. is made conventionally with absorption train while the cartridge of MgO is treated with steam at 1000°. The resulting HF is detd. by titration. The MgO can be reused after drying. If the sample contains elements such as P, B, or Si which are also retained by MgO, the detn. is unaltered, but in presence of halogens or S which form compds. with Mg that are hydrolyzable by steam, the titration of HF is possible only with Th nitrate. MgO retains S completely, Cl and Br only partially. A set of typical analyses are shown indicating the accuracy of about 0.2% or better for C, 0.1% for H, and 0.1-0.2% for F. G. M. Konstantinoff

pro. RM orig



GEL'MAN, N. E.

✓ 143. Micro- and semi-micro determination of nitrogen by hydrogenation of organic materials. N. E. Gel'man and M. O. Korshun Izv. Akad. Nauk SSSR, Ser. Khim., 1957, 12 (1), 123-125.—The method is based on a rapid thermal decomposition of the material in a current of H<sub>2</sub> and passage of the products over an iron catalyst prepared by igniting Fe<sub>2</sub>(NO<sub>3</sub>)<sub>6</sub> and reducing the oxide in H<sub>2</sub>. The NH<sub>3</sub> formed from the N in the material is absorbed in 0.02 N KH(IO<sub>3</sub>)<sub>2</sub>, the excess of which is determined iodimetrically. The method is applicable to the analysis of amino compounds and heterocyclic compounds containing C, H, O, S and S.

452  
454

NS // 172

GEL'MAN, H. Ye.; KORSHUN, M. O.; SHEVELEVA, N. S.

Rapid methods of elementary microanalysis. Report No. 14:  
Determining microquantities of carbon and hydrogen in fluorine  
organic compounds [with summary in English]. Zhur.anal.khim.  
12 no.4:526-533 J1-Ag '57. (MIRA 10:10)

1. Institut elementoorganicheskikh soedineniy AN SSSR, Moskva.  
(Carbon) (Hydrogen) (Fluorine organic compounds)

5(3)

AUTHORS:

Korshun, M. O., Gol'man, N. E.,  
Sheveleva, N. S.

SOV/75-13-6-16/21

TITLE:

Rapid Methods of Micro-Elementary Analysis (Skorostnyye metody mikroelementarnogo analiza) Communication 15. On the Problem of Simultaneous Micro-Determination of Carbon, Hydrogen, and Halogens in Organic Compounds (Sobshcheniye 15. K voprosu ob odnovremennom mikroopredelenii ugleroda, vodoroda i galoidov v organicheskikh soyedineniyakh)

PERIODICAL:

Zhurnal analiticheskoy khimii, 1958, Vol 13, Nr 6, pp 695-701 (USSR)

ABSTRACT:

It was established in earlier papers (Refs 1-4) that the presence of halogens or of compounds containing halogens in the combustion of organic substances hinders oxidation of carbon to CO<sub>2</sub>. Therefore, a platinum contact must be used in this case for the quantitative oxidation. Recently, the authors developed a method for burning in a so-called "kull" which allows to improve the determination of C, H and halogens considerably (Ref 5). A new variant of this method is described in the present paper. In the vessel containing the weighed portion, silver, and the

Card 1/4

Rapid Methods of Micro-Elementary Analysis.

SOV/75-13-6-16/21

Communication 15. On the Problem of Simultaneous Micro-Determination of Carbon, Hydrogen, and Halogens in Organic Compounds

oxidation zone follow one another in the combustion tube in the direction of the gas current. Both silver and the vessel containing the weighed portion are placed in a thin hollow quartz hull, which is weighed out after combustion. In this case no platinum contact is required in the oxidation zone (Ref 4) and the adoption of the hull allows the silver to be weighed out. The hull weighs about half of the former massive appliance and therefore secures a far better reproducibility of halogen determination. No combustion tubes with ground apparatus are required any longer. Tube life is also prolonged, as it cannot be corroded by the silver contained in the hull. Pure metallic silver in the form of a foil, a net or a wire is used for the absorption of halogens. Only 1.5 g Ag are required, which is much less than the formerly used appliance called for. The silver layer is heated to 550-600° by means of a MAG-6R electric burner. From 30 to 40 determinations can be carried out with the used amount of silver. A temperature increase from 425 to 575° causes the absorbability of silver to increase considerably. In the case of temperatures being low to an extent at which it is

Card 2/4

Rapid Methods of Micro-Elementary Analysis.

SOV/75-13-6-16/21

Communication 15. On the Problem of Simultaneous Micro-Determination of Carbon, Hydrogen, and Halogens in Organic Compounds

not possible to work with an Ag net or foil, other large surface silver preparations have a good absorbability. In this connection the authors investigated silvered pumice, as silver deposited upon a porous carrier efficiently absorbs halogens and corrodes the quartz hull much less than metallic silver. A granulated, silvered pumice prepared according to Sokolova's method (Ref 7) was highly suitable for the determination. Absorbability of this preparation is almost twice that of electrolytic silver (Ref 6). Halogen absorption may thus be carried out at 425° instead of at 575°, in which connection corrosion on the hull is so slight that it can be repeatedly used again. Carrying out of this new determination method as well as the results of the several analyses are accurately described. This method can also be used for the determination of C, H, and S and for that of some other elements from a weighed portion. There are 3 figures, 7 tables, and 14 references, 13 of which are Soviet.

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Rapid Methods of Micro-Elementary Analysis.  
Communication 15. On the Problem of Simultaneous Micro-Determination of Carbon,  
Hydrogen, and Halogens in Organic Compounds

SOV/75-13-6-16/21

ASSOCIATION: Institut elementoorganicheskikh soyedineniy AN SSSR, Moskva  
(Institute of Organic Elemental Compounds of the Academy of  
Sciences, USSR, Moscow)

SUBMITTED: September 12, 1957

Card 4/4

5(2,3)

AUTHORS:

Gel'man, N. E., Korshun, M. O.,  
Chumachenko, M. N., Larina, N. I.

SOV/20-123-3-24/54

TITLE:

Analysis of Organofluoric Compounds (Analiz ftororganicheskikh  
soyedineniy )Simultaneous Micro-Determination of Fluorine and  
Nitrogen in Organic Compounds (Odnovremennoye mikroopredeleniye  
ftora i azota v organicheskikh soyedineniyakh)

PERIODICAL:

Doklady Akademii nauk SSSR, 1958, Vol 123, Nr 3, pp 468-470  
(USSR)

ABSTRACT:

In previous papers by the authors (Refs 1, 2) it was found that  
magnesium oxide in the elementary analysis of organofluoric  
compounds is a reliable reagent for a quantitative linkage of  
fluorine which is separated out in decomposition of organic  
substances. Moreover, they proved that in fluorine, absorbed by  
MgO, can be quantitatively isolated from the absorbing layer as  
HF by the hydrolytic decomposition of magnesium fluorine by  
vapor at a high temperature (Ref 3). This so-called  
pyrohydrolysis proceeds as follows:  $MeF + H_2O \rightarrow MeO + HF$ . On

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account of this result, an experiment was carried out with the  
process mentioned in the subtitle. For this purpose the

Analysis of Organofluoric Compounds. Simultaneous SOV/20-123-3-24/54  
Micro-Determination of Fluorine and Nitrogen in Organic Compounds

modification of nitrogen-determination by Dumas (Lyuma) was used, which had been worked out by the second and the third authors. In this process the measured amount was burned by means of pyrolysis in a layer of nickel oxide. Nickel oxide did not disturb the pyrohydrolytic determination of fluorine (Ref 5). Table 1 shows the results of the determination of fluorine by combustion at  $900^{\circ}$  -  $950^{\circ}$  in a  $\text{CO}_2$  atmosphere in an electric furnace (length: 6 cm). 3-8 mg of the substance were used, which were covered by a layer of granulated nickel oxide in a quartz tube. For the pyrohydrolysis a tube was used that had been suggested by N. E. Gel'man and K. I. Glazova. The pyrohydrolysis takes 20-25 minutes. Accuracy of the determination: nitrogen 0.2%, fluorine up to 0.5% absolute. The results are shown in table 2. The authors were the first to carry through this determination. There are 2 tables and 6 references, 5 of which are Soviet.

ASSOCIATION: Institut elementoorganicheskikh soedineniy Akademii nauk SSSR  
(Institute of Elemento-Organic Compounds of the Academy of Sciences, USSR)

Card 2/3



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**252**

# Division of Analytical Chemistry of the VIII Mendeleev Congress on General and Applied Chemistry

## PHYSIOLOGICAL

**Journal of Microbiology** 1979, Vol 14, No 4, pp 511-512  
(1979)

## APPENDIX

**Case 7/0**

**Card 2/4**

case 5/4

[illegible]

3. N<sup>1</sup> N<sup>2</sup> W 739.

U.5230

7/1/51  
SC7/75-15-1-15/29

AUTHORS: Korshun, M. O. (deceased), Sheveleva, N. S., Gel'man, N. E.

TITLE: Rapid Methods of Microanalysis. Communication 17. Simultaneous Determination of Carbon, Hydrogen, Mercury, and Halogen in a Single Sample of Organic Substance

PERIODICAL: Zhurnal analiticheskoy khimii, 1960, Vol 15, Nr 1, pp 99-103 (USSR)

ABSTRACT: The article describes determination of mercury and halogen (or sulfur) since carbon and hydrogen are determined by the usual methods. An organic sample is subjected to rapid pyrolytic combustion in which the resulting mercury vapors are retained by gold (thin wire or foil) and halogen by silver (screen, wire, or foil). Silvercoated pumice cannot be used since it retains part of the mercury and distorts the experimental results. After the retention, mercury and halogen are determined gravimetrically. The experimental error is not more

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Rapid Methods of Microanalysis. Communication  
II.

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SOV/15-15-1-15/25

than 0.6 to 0.7% for mercury, and less for halogen.  
It was found that mercury can also be determined in  
substances containing nitrogen, since no mercury  
nitrate is formed during the rapid combustion. There  
are 3 tables; 4 figures; and 13 references, 4 German,  
8 Soviet, 1 U.K. The U.K. reference is: Heron, A. E.,  
Analyst, 72, 142 (1947).

ASSOCIATION: Institute of Element-Organic Compounds, Academy of  
Sciences, USSR, Moscow (Institut elementoorganicheskikh  
soyedineniy AN SSSR, Moskva)

SUBMITTED: December 13, 1958

Card 2/2

GEL'MAN, N.E.; KORSHUN, M.O. [deceased]; NOVOZHILOVA, K.I.

Analysis of fluorine organic compounds; use of pyrohydrolysis for the simultaneous micro determination of fluorine, carbon, and hydrogen. Zhur.anal.khim. 15 no.2:222-226 Mr-Apr '60.

(MIRA 13:7)

1. Institut elementoorganicheskikh soyedineniy AN SSSR, Moskva.

(Fluorine organic compounds)

(Fluorine--Analysis)

(Carbon--Analysis)

(Hydrogen--Analysis)

GEL'MAN, N.E.; KORSHUN, M.O. [deceased]; NOVOZHILOVA, K.I.

Analysis of fluoroorganic compounds. Simultaneous microdetermination of fluorine, carbon, and hydrogen. Zhur.anal.khim.  
15 no.3:342-346 My-Je '60. (MIEA 13:7)

1. Institute of Elemento-Organic Compounds, Academy of Sciences,  
U.S.S.R., Moscow.

(Fluorine--Analysis) (Carbon--Analysis)  
(Hydrogen--Analysis)

S/075/60/015/004/022/030/XX  
B020/B064

AUTHORS: Gel'man, N. E. and Van Ven'-yun'

TITLE: Conductometric Microdetermination of Carbon and Hydrogen in Organic Compounds \

PERIODICAL: Zhurnal analiticheskoy khimii, 1960, Vol. 15, No. 4,  
pp. 487 - 494

TEXT: In working out a method for the conductometric determination of the combustion products of carbon and hydrogen, the existing variants of the carbon determination method (at a low carbon content) in organic substances were adapted to an accuracy of 0.2 - 1% by a cell for measuring the electrical conductivity. The basis for the conductometric determination of hydrogen was its reaction with carbon at high temperatures in an inert medium; in the present case it served to determine the hydrogen and carbon in the organic substance from one weighed portion. The determination is carried out in two stages: the organic substance is first burned in an empty tube in the oxygen current, the resulting water obtained is frozen out with a dry ice - acetone mixture, and CO<sub>2</sub> introduced with dilute lye

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Conductometric Microdetermination of Carbon  
and Hydrogen in Organic CompoundsS/075/60/015/004/022/030/XX  
B020/B064

into the cell. The frozen water is passed in nitrogen or argon current over a red-hot layer of platinized carbon black, where the oxygen of the water reacts with carbon under the formation of CO, which oxidizes to CO<sub>2</sub> over heated Cu<sub>2</sub>O, and is also passed into the cell. Both carbon and hydrogen are determined by the change of conductivity of the absorption solution. The reproducibility of conductometric carbon and hydrogen determination is better than that of their gravimetric determination. Moreover, this method permits to follow the combustion process of the weighed portion in time, and automate the measurement of conductivity. The suggestion is made to carry out the combustion process automatically at high temperatures and great oxygen excess. 65 organic compounds were analyzed by this method (cf. Table 1). The apparatus used to burn and determine carbon (Fig. 1), and to convert water (Fig. 2) are described. The device used to determine the electrical conductivity is designed according to the Wheatstone bridge, and consists of a 3P-10 (ZG-10) sound generator as a.c. source, an MTB (MTV) d.c. bridge, three resistors as arms of the Wheatstone bridge, an ME-3<sup>2</sup> (MYe-3) capacitor for compensation and self-induction, an EO-7 (EO-7) electron oscillograph as zero

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Conductometric Microdetermination of Carbon  
and Hydrogen in Organic Compounds

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B020/B064

instrument, and two cells to measure the electrical conductivity, which were modified in accordance with the requirements made (Fig. 3). The results of checking the quantitative  $\text{CO}_2$  absorption in the measuring cell are listed (Table 2). Platinized carbon black with a 50% Pt content was obtained by a method of Ye. A. Bondarevskaya (Ref. 23). The weighing in, the course of analysis when a 0.01 N  $\text{Ba}(\text{OH})_2$  solution is used and a

0.01 N NaOH solution as absorbent are described; then, a practical example is calculated. The authors thank N. A. Balashova for valuable advice and interest in the work, and Yu. S. Solomatin for mounting the electrical measuring apparatus and designing the automatic rotary furnace. There are 3 figures, 2 tables, and 30 references: 17 Soviet, 7 German, 1 French, 3 Austrian, and 2 US. ✓

ASSOCIATION: Institut elementoorganicheskikh soedineniy AN SSSR, Moskva  
(Institute of Elemental-organic Compounds of the AS USSR,  
Moscow)

SUBMITTED: January 25, 1960

Card 3/3



05030

3/075/60/015/005/024/026/XX  
B002/B056

11. 2214  
AUTHORS: Gel'man, N. E., Korshun, M. O. (Deceased), and Novozhilova, K. I.

TITLE: The Analysis of Organofluorine Compounds.<sup>1</sup> The Simultaneous Micro Determination of Fluorine, Carbon and Hydrogen in Low-boiling and Gaseous Compounds

PERIODICAL: Zhurnal analiticheskoy khimii, 1960, Vol. 15, No. 5, pp. 628-634

TEXT: The essential difficulty in the analysis of low-boiling compounds consists in the fact of having to determine an exactly weighed substance and conveying it without losses into the combustion tube. The authors show that substances with a boiling point of 20°C may still be weighed in an open quartz capillary - diameter of the opening from 0.2 to 0.3 mm ; the losses e.g. at 3,3,3,2,1-pentafluoropropane (boiling point 20°C) amount to 0.024 mg per minute with a weighed portion of 4 mg. If the evaporation rate is known, the losses may be corrected in the time between

Card 1/3

85638

The Analysis of Organofluorine Compounds. The S/075/60/015/005/024/026/XX  
Simultaneous Micro Determination of B002/B056  
Fluorine, Carbon and Hydrogen in Low-boiling and Gaseous Compounds

the weighing and conveying into the combustion tube. The following substances were determined according to this method: 1-ethylfluoroisobutylene  $C_6H_5F_7$ ; 1-bromine-2-hydroperfluoro-isobutane  $C_4HBrF_8$ ; 3,3,3,1-tetrafluoro-2-trifluoromethylpropane  $C_4H_3F_7$ ; 1,2-dihydroperfluoroisobutane  $C_4H_2F_8$ ; 3,3,3,2,1-pentafluoro propane  $C_4H_3F_5$ . For substances with a lower boiling point (between  $+20^{\circ}$  and  $-45^{\circ}C$ ), an improved "opener" according to Yu. N. Dagayeva and K. I. Novozhilova was used. The substance is conveyed into a bulged capillary, whose outer walls are cooled by means of a freezing mixture. The capillary is sealed, weighed, and broken in the combustion tube by opening the oxygen faucet. The following substances were determined by means of this method: monohydroperfluoroisobutylene  $C_4HF_7$ ; monohydroperfluoroisobutane  $C_4HF_9$ ; 3,3,3,2,1-hexafluoropropane  $C_3H_2F_6$ ; monohydroperfluoropropane  $C_3HF_7$ ; 3,3,3,2-pentafluoropropene-1

Card 2/3

85638

The Analysis of Organofluorine Compounds. The  
Simultaneous Micro Determination of  
Fluorine, Carbon and Hydrogen in Low-  
boiling and Gaseous Compounds

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B002/B056

$C_3HF_5$ ; hexafluoroacoxymethane  $C_2F_6N_2O$ ; trifluoronitromethane  $CF_3NO_2$ ;  
difluorochloromethane  $CHClF_2$ ; 3,3,3-trifluoropropene-1;  $C_3HF_3$ ;  
1,1,2-trifluoroethene  $C_2HF_3$ . For substances having a lower boiling point  
than  $-45^\circ C$ , an ordinary gas burette is used, by means of which the  
following substances were once again determined: hexafluoroacoxymethane,  
trifluoronitromethane, difluorochloromethane, 3,3,3-trifluoropropene-1,  
1,1,2-trifluoroethene. There are 6 figures, 3 tables, and 18 references:  
5 Soviet, 7 US, 5 British, and 1 German.

ASSOCIATION: Institut elementoorganicheskikh soedineniy AN SSSR, Moskva  
(Institute of Elemental-organic Compounds AS USSR, Moscow)

SUBMITTED: April 13, 1959

Card 3/3

TERENT'YEV, A.P., otv.red.; ALIMARIN, I.P., red.; GEL'MAN, N.E., red.;  
KLIMOVA, V.A., red.; KRISHKOV, A.P., red.; KUZNETSOV, V.I., red.;  
LEVIN, E.S., red.; PODGAYSKAYA, Z.I., red.; RUKHADZE, Ye.O., red.;  
TAL'ROZE, V.L., red.; TSUCKERMAN, A.M., red.; SHERYAKIN, F.M., red.;  
SHOYINKER, Yu.N., red.; YERMAKOV, M.S., tekhn.red.

[Conference on organic analysis] Soveshchanie po organicheskomu  
analizu. Tezisy dokladov. Moskva, Izd-vo Mosk.univ., 1961. 170 p.  
(MIRA 14:4)

1. Soveshchaniye po organicheskomu analizu. 1961.  
(Chemistry, Analytical--Congresses)  
(Chemistry, Organic--Congresses)

S/032/61/027;001/005/037  
B017/B054

AUTHORS: Gel'man, N. E., Van Ven'-yun', and Bryushkova, I. I.

TITLE: Use of Conductometry for a Direct Microdetermination of  
Oxygen in Organic Compounds

PERIODICAL: Zavodskaya laboratoriya, 1961, Vol. 27, No. 1, pp. 24-28

TEXT: A direct conductometric microdetermination of oxygen was developed according to the method by M. O. Korshun and Ye. A. Bondarevskaya (Refs. 10, 11). The organic compound is thermally decomposed in a nitrogen- or argon atmosphere; the resulting gaseous reaction products are allowed to pass over platinized carbon black at 900°C, the oxygen is quantitatively transformed to CO. The resulting carbon monoxide is oxidized by copper monoxide to CO<sub>2</sub> at 300°C, and is absorbed in an alkaline solution. The resulting carbon dioxide is determined by the change in electrical conductivity of the absorption solution. For a quantitative oxidation, a 3.5 cm long contact layer is required, and the gas flow velocity must not exceed 10-12 ml/min. Numerous organic compounds of

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Use of Conductometry for a Direct Micro-  
determination of Oxygen in Organic Compounds

S/032/61/027/001/005/037  
B017/B054

different structures and compositions were analyzed; results are compiled in Tables 1 and 2. An analysis takes 30-35 minutes. The percent oxygen content in organic compounds was determined by a calibration curve shown in Fig. 3. The oxygen amount in  $\gamma$  is plotted on the abscissa, the decrease in electrical conductivity of the absorption solution on the ordinates. There are 3 figures, 2 tables, and 11 references: 6 Soviet and 5 German.

ASSOCIATION: Institut elementoorganicheskikh soedineniy Akademii nauk  
SSSR (Institute of Elemental-organic Compounds, Academy of  
Sciences USSR)

Card 2/2

GEL'MAN, N.E.; SHANINA, T.M.

Quantitative analysis of heteroorganic compounds. Microdetermination of phosphorus. Zhur.anal.khim. 17 no.8:998-1004 N '62. (MIRA 15:12)

1. Institute of Heteroorganic Compounds, Academy of Sciences, U.S.S.R., Moscow.

(Phosphorus—Analysis)

(Phosphorus organic compounds)

GEL'MAN, N.E.; BRYUSHKOVA, I.I.

Elemental analysis of organometallic compounds igniting in air. Simultaneous microdetermination of carbon, hydrogen, and aluminum or some other element as an oxide. Zhur. anal. khim. 19 no.3:369-374 '64. (MIRA 17:9)

1. Institut elementoorganicheskikh soyedineniy AN SSSR, Moskva.



GEL'MAN, N.B.; LARINA, N.I.

Elemental analysis of organofluorine compounds. Simultaneous  
microdetermination of fluorine, chlorine, and nitrogen. Zhur,  
anal. khim. 19 no.5:593-597 '64. (MIRA 17:8)

1. Institut elementoorganicheskikh soyedineniy AN SSSR, Moskva.

SHANINA, T.M.; GEL'MAN, N.E.; KIPARENKO, L.M.

Quantitative analysis of organometallic compounds. Spectro-  
photometric microdetermination of silicon. Zhur. anal. khim. 20  
no.1:118-125 '65. (MIRA 18:3)

1. Institut elementoorganicheskikh soyed:neniy AN SSSR, Moskva.

GEL'MAN, N.B.; SHEVELEVA, N.S.

Quantitative elementary analysis of organic compounds. Micro-  
determination of carbon and hydrogen by burning in a micro-  
test tube. Zhur. anal. khim. 20 no.6:719-726 1964. (USSR 1964)

1. Institut elementoorganicheskikh soedineniy Akad. Nauk SSSR, Moscow.

GEL'MAN, N.E.; BPESLER, P.I.; RUZIN, B.N.; GREK, N.V.; SHEVELEVA, N.S.;  
MELNIKOVA, A.A.

New method for the automatic microdetermination of carbon and  
hydrogen in organic compounds. Dokl. AN SSSR 161 no.1:107-110  
Mr '65. (MIRA 18:3)

1. Institut elementoorganicheskikh soedineniy AN SSSR i Spetsial'-  
noye konstruktorskoye byuro analiticheskogo priborostroyeniya AN  
SSSR. Submitted July 29, 1964.

GEL'MAN, N.F.

Fine planing instead of scraping. Stan.1 instr. 2<sup>1</sup>/<sub>4</sub> no.10:24-25 0 '53.  
(MIRA 6:11)  
(Planing machines)

GEL'MAN, N.L., inzhener.

~~XXXXXXXXXXXX~~

Two cases of switching on moist generators. Elektrichestvo no.11:70 N '53.  
(MLRA 6:10)

1. Rostovenergo.

(Dynamos)

GEL'MAN, N. L.

AID P - 3526

Subject : USSR/Power Eng  
Card 1/1 Pub. 26 - 20/30  
Author : Gel'man, N. L., Eng.  
Title : Decreasing the loss angle of oil without regeneration  
Periodical : Elek. sta., 9, 55, S 1955  
Abstract : The article describes the installation of a transformer with defective insulation the loss angle of oil became too great. A thermosyphon was installed and after 6 days the loss angle of oil was brought down to normal.  
Institution : None  
Submitted : No date

KURTSVAYL', G.I., inzhener; GEL'MAN, N.L., inzhener; DONIK, A.N., inzhener.

Defects of FS-600 circuit breakers. Energetik 4 no.8:9-10 Ag '56.  
(Electric circuit breakers) (MIRA 9:10)



ORL'MAN, N.L., inzhener.

Simple device for measuring the distance between conductors at intersections of overhead lines. Energetik 4 no.10:27-28 0 '56.  
(MLBA 9:11)

(Electric lines--Overhead)  
(Optical instruments)

GEL'MAN, N. L., inshener.

Repairing a generator following breakdown of the stator winding  
during testing. Elek.sta.27 no.12:49-50 D '56. (MLRA 10:1)  
(Electric generators)

GEL'MAN, N.L., inshonor: KURTSVAYL', G.I., inshonor.

Case of damage to a KRU cell with a VMO-133-II cutout. Elek.  
sta. 28 no.5:75-76 My '57. (MLRA 10:6)  
(Electric transformers)

GEL'MAN, N.L., inzh.

Placing a damaged 20,000 kva. transformer into operation. Elek.  
sta. 32 no.2:76-77 P '61. (MIRA 16:7)  
(Electric transformers)  
(Electric power distribution)